



Synthesis of Polymerization of New Derivative of 2-Decanoic Acid Extracted from Coriander Seed

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Abstract

This work, includes preparation of the derivative of decanoic acid was extracted from conriander seeds by fractional distillation for four hours, then reacted with one groups of amin of thio urea the 2nd group amin in thiourea was reacted with maleic anhydride a different drugs was used to obtain novel derivatives. The compounds written off as by using some spectroscopic methods, H¹NMR and FT-IR used to characterization the derivatives, control drug release was studying of some derivatives, biological activity against bacterial was studying and thermal stability of the derivatives.

Keywords: Decanoic acid, Conriander seeds, FT-IR, H1NMR

Introduction

The important oils and excerpts of perfumed vegetation and spices have been used in nutrition conservation, pharmaceuticals products, stand by medicine and natural treatments.[1] Currently, Those plants need to be verified scientifically for the configuration of essential oil and its biological activities, to improve the quality of healthcare so which have used in classical medicine[2]. The oil insides in dissimilar types are varied inherently, affected greatly on culture conditions and environment, as well as yield and after harvesting, and hence assessments of the oils from many medicinal plants are being conducted.

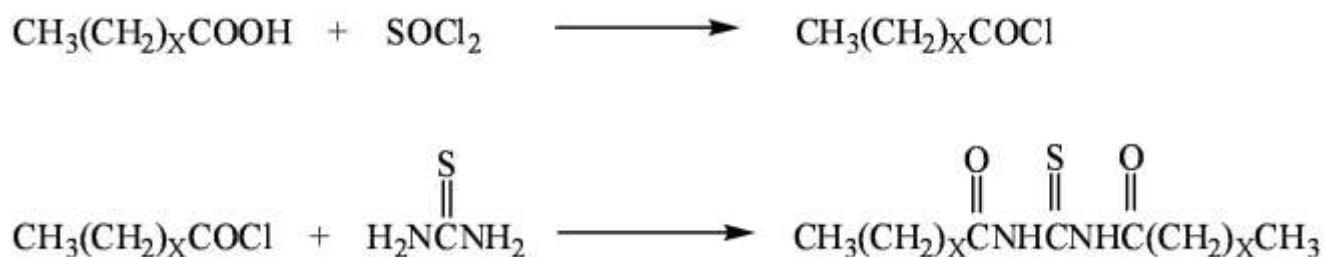
One of the most useful EO bearing spices as well as medical plants is Coriandrum sativum [3]. Coriander an annual herb of the parsley family detection of α -pinene, limonene, β -.The organic conformation of foils and grain fats of C. Sativum. This countryside approval it to be recognized for therapeutic practice and off the record amongst the further oils existing in the worldwide flea market [4].Examined of some chemicals and biological compounds which have bioactive caustic. Newly, several widespread assessments of coriander have

issued, focusing only going on their biologically active modules, mainly phenolic compounds, In addition to many medical accomplishments. [5-6] Coriander roots have flavor the deeper leaves are usually consumed in many gastronomies in Asian regions, while the coriander is cut besides secondhand in soup and flour [7-8]. It has proved its taste and aromatic properties, coriander, it is essential pharmaceutical parsley as point out via a number of spice dealer. Coriander is present in curatives preparations beside gastrointestinal troubles in stock treatment.

In the north of Pakistan, full coriander leaves are popularly used for abdominal swelling, diarrhoea, purge, coughing, grumbles of stomach, jaundice and puke [9-10].In olden India medicine, coriander was used to treat joint problems and demagogic illnesses. In fact, classical polyhedral formulation. Differences in the organic appearance of coriander in needed oils can be understood from corner to corner areas taking about description whole reasons effecting the organic structure of oils, ie, inborn, climatological, seasonal, geographic too other situations [11-12].

Decanoic acid, a 10 carbon containing carboxylic acid), is an important precursor in the production of nylon and polyamides the commercial process for the production of Decanoic acid depends on chemical methods, principally involving alkaline oxidation of vegetable oils, such as castor oil, in which 2-octanol is generated as a byproduct [13-14]. However, the production of medium-chain through chemical routes has several problems, particularly the use of harsh production conditions and generation of by-products. Biological processes for the

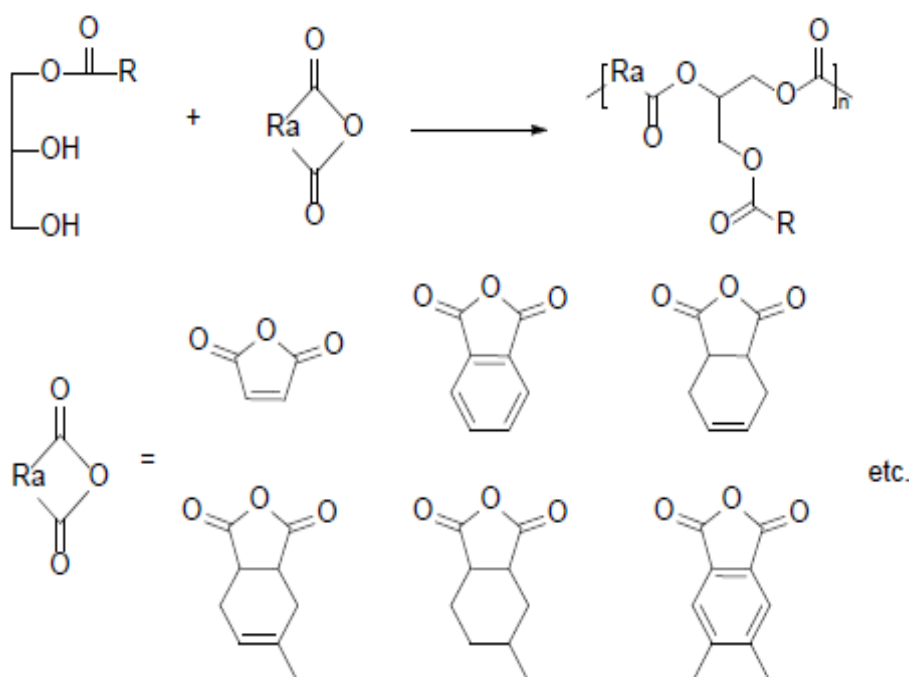
production of DCAs can overcome these limitations microorganisms, such as *E. coli*, has been a focus of several studies [15-16] the toxicity of decanoic acid is considered to be a major limitation in the biological production Thiourea, having a considerably wide range of applications, is a functional organic compound similar to urea, except that the oxygen atom is replaced by a sulfur atom. The properties of urea and thiourea differ significantly because of the difference in electronegativity between sulfur and oxygen.



Scheme 1: general reaction of acids with thiourea

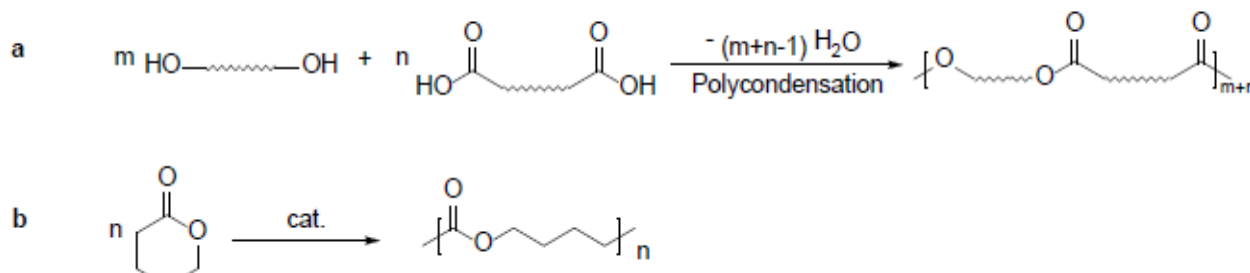
ω -Hydroxy fatty acids are a class of straight long-chain aliphatic organic compounds with carboxylic acid and hydroxyl functionalities located at both ends of the fatty acid chain. The ω -hydroxy fatty acids have been attracted much attention for preparation of mixed diesters, cosmetic formulations and phospholipids [17] Polyesters are the most widespread used biodegradable polymeric materials for drug carrier and tissue engineering. Polyesters can be synthesized either by ring opening polymerization (ROP)

of cyclic ester monomers or polycondensation of two multifunctional monomers. Condensation polymerization is a polymerization process by which two molecules (monomer, oligomer or polymer molecules) are linked together by generating a new functional group with elimination of a small molecule resulting elongation of polymer chains Vegetable oil based polyester resins (alkyds) were mainly obtained by polycondensation from monoglyceride with dicarboxylic acid anhydride [18-19].



The properties of the polyesters can be adjusted by using different type of dicarboxylic anhydride and monoglyceride/anhydride feed ratios. When dicarboxylic acid anhydrides with rigid structures are used, polyesters with higher glass transition temperatures and cross-linking densities will be obtained Polyester can be synthesized by condensation polymerization with the elimination of water molecules in each condensing [20-21]). ROP of cyclic ester monomers is one of the most

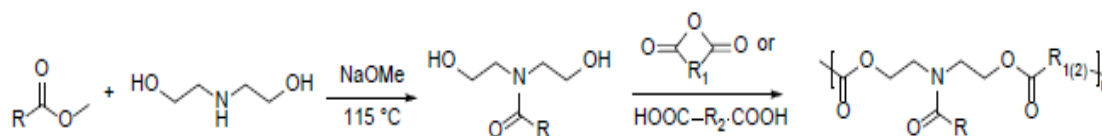
effective methods to obtain homo- or copolyesters with a high molar mass and low dispersity under milder reaction conditions in comparison to condensation polymerization [22-23]. Metallic compounds [61], Guanidine based catalysts [24-25], strong acids [26-27], phosphazene [28], N-heterocyclic carbene [29, 30]. Bifunctional thiourea-amine [31-32] and enzyme [33-34] have been applied as catalysts to the controlled ROP of cyclic esters.



Scheme 2.a: Typical polycondensation of synthesis of polyester by ring-opening polymerization of a lactone

The synthesis of vegetable oil based polyesteramide is generally accomplished by condensation polymerization of N, N-bis (2-hydroxy ethyl) fatty amides which were

obtained by amidation of fatty acids with diethanolamine, with various dibasic acid or anhydrides [35-36].



R = fatty acid chain

Scheme 3: Synthetic route of vegetable oil based polyesteramide

The aim of this study was to synthesize and characterize fatty acid-grafted-maleic anhydrid (fatty acid-g-MA) polymer and loaded for use as carriers for drug delivery.

Experimental

Materials and Instruments

Decanoic acid was obtained from Coriandrum seed. Ethanol and acetone remained obtained as of Merck. ¹H-NMR spectra documented by spectrophotometer of a Shimadzu in Dimethylsulphoxide (DMSO). The FTIR spectra documented by (4000-400cm⁻¹) at a Shimadzu

Synthetic Methods of Compounds

- 5 gm of coriander seed was dissolved (25 ml) in mixture of ethanol and acetone.

- Distillation solution about 3hr (70C°). Until the disappearance of oily substance.
- He extracted substance (decanoic acid) is collected and divided to part the first part was added few drops of thionyl chloride and then reacted with 1 mol of thiourea.
- The mixture was refluxed about 2hr, the product was collected and washed with diethyl ether solvent.

Paracetamol dissolved in DMF and reacted with the mixture, and refluxed about for 1hr, the result was swashed in ethanol. Then we let it dry the room warmly .another drug was prepared in the same procedure.

The second part of extracted reacted with maleic anhydride and then refluxed about one hour, few drops of ammonium persulphat APS added to the mixture and heated 5mints

viscous yellow product was obtained ,different amino drugs was reacted with the mixture

Result and Discussion

The main objectives of the present research were the synthesis of new derivatives of Decanoic [1] by reaction of decanoic acid with thiourea and maleic anhydride in presence of with acetone as a solvent The FTIR spectra of prepared compound(1,2,3) confirmed by the peaks at (3178,3180,,3190 cm-1) due to N-H stretching. totaling to the overhead point out peaks, the peaks seeming in the assortment (3000 to 2900) cm-1 appearances the attendance of (-CH₂) group in completely three spectra. Apart as of the extra ordinarily detected peaks spectra are (1703 cm-1carbonyl group (C=O) 1H-NMR spectra of compound [9] δ ppm presented singlet signal of ($\delta = 5.5, 4$) ppm owing to (, H1, N-H) proton, singlet signal $\delta = (4, 1-)$ ppm owing to (3H, CH₃), another data was tabulated in the tables 1and 2.

Thermogravimetric Analysis (TGA) measures the amount and rate of change in the weight of a material as a function of temperature or time in a controlled atmosphere. TGA is a useful technique to assess the thermal stability of polymer. It is an important technique in which the mass of the substance is measured as a function of temperature, while the substance is subjected to controlled temperature programme. In any polymer analysis, the TG trace follows a relatively simple pattern. Thus the sample weight decreases slowly as the reaction begins, then decreases rapidly over a corresponding narrow temperature range and finally comes to a minimum as the reactants are spent.

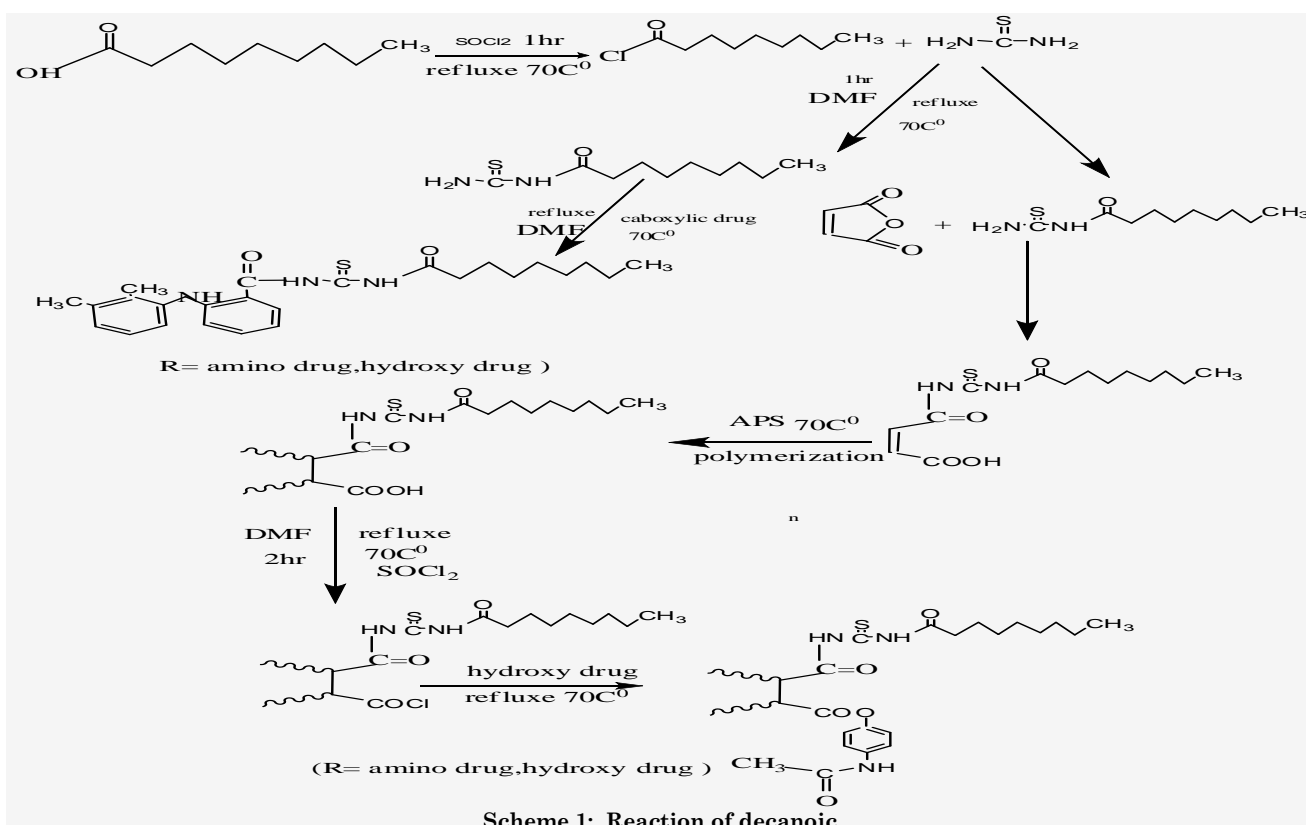
The conventional controlled release dosage form is the inability to increase its residence time for example in pH 1.1 of small intestine, resulting in an improved and bioavailability of the basic drug and to prolong the presences of dosage form in the stomach until all the drug is released in the desired period of time.

Table1: FT-IR of compounds (1)

Com	N-H	C-Harm	C-H alp	C=O	C=C
1	3178	3022	2989	1703	1654
2	3190	3008	2927	1745	1622
3	3180	3068	2929	1743	1653

Table 2: Hnmr-Spectral of Compound

Com	N-H	CH ₃	CH ₂	CH=CH	PH	CH-N
1	5.5	4	4-5	4	6-7	5
2	4	1	2,2,5,3	3,5	6-7	5



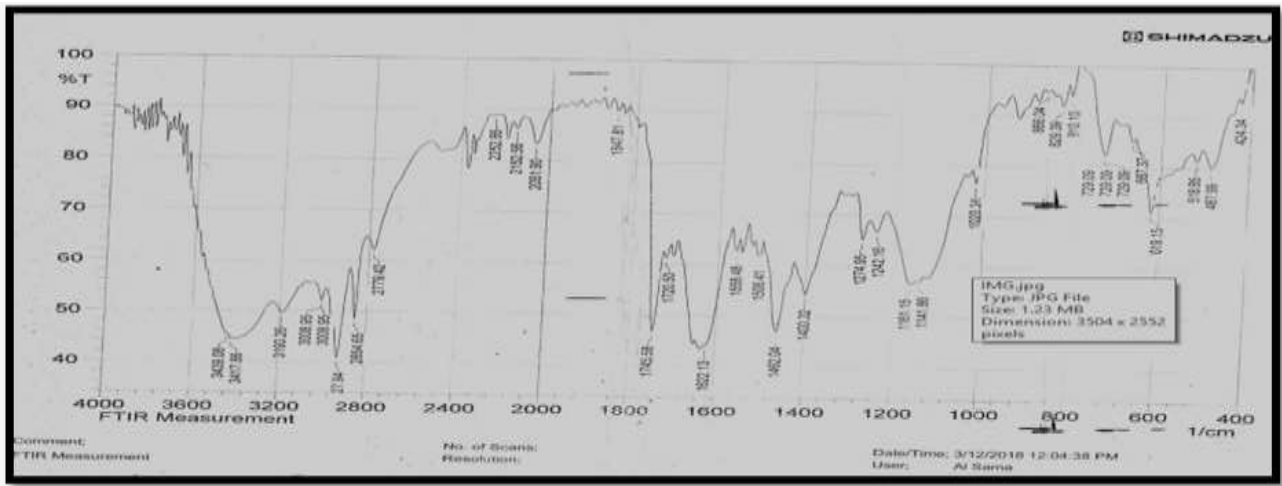


Figure 1: FTIR of decanoic acid with thiourea

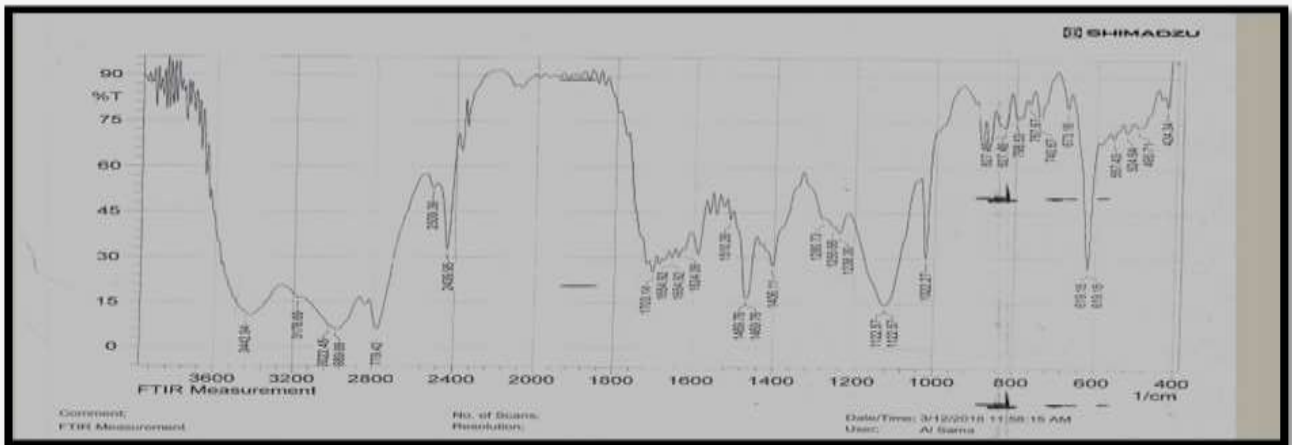


Figure 2: FTIR of decanoic acid with thiourea with amoxicillin

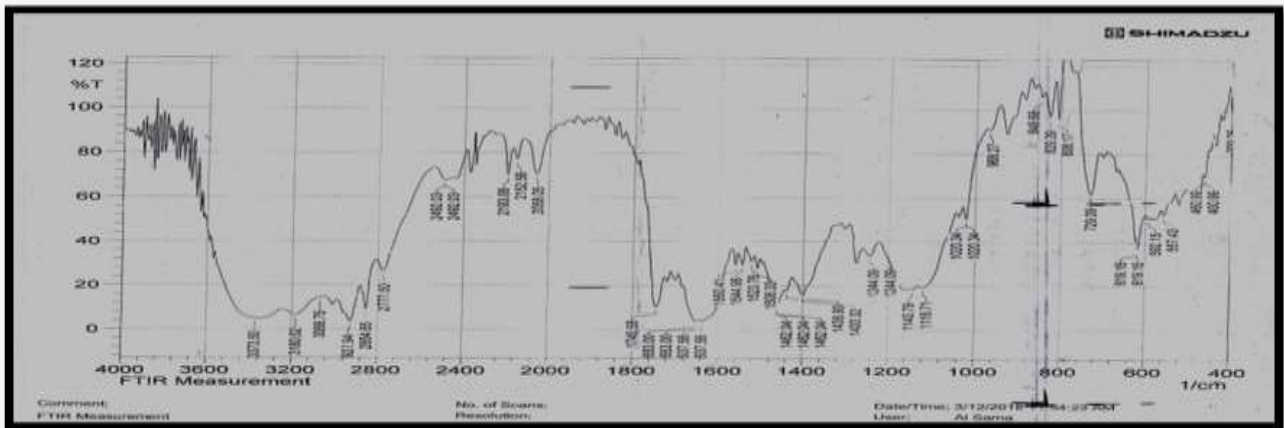


Figure 3: FTIR of decanoic acid with thiourea with paracetamol

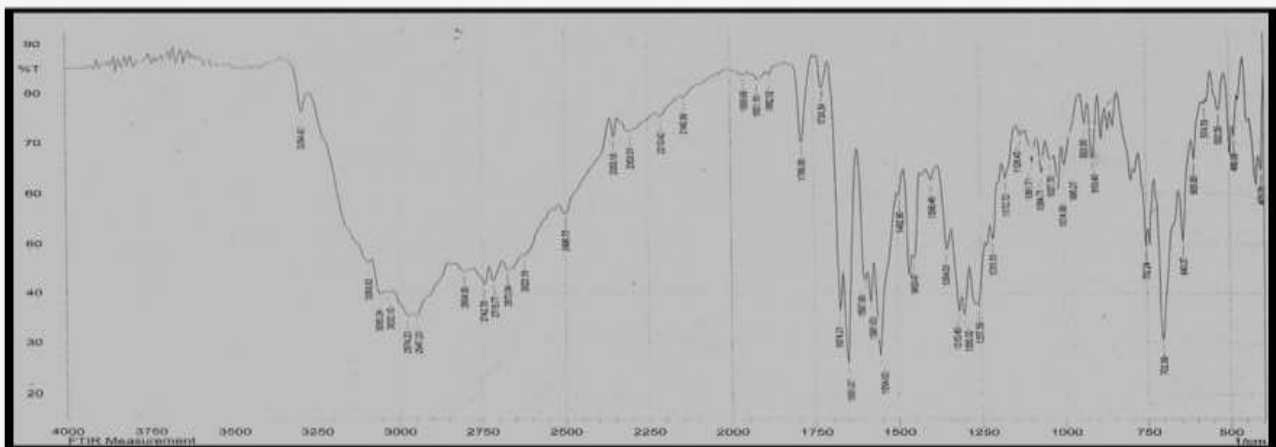


Figure 4: FTIR of decanoic acid with maleic anhydride

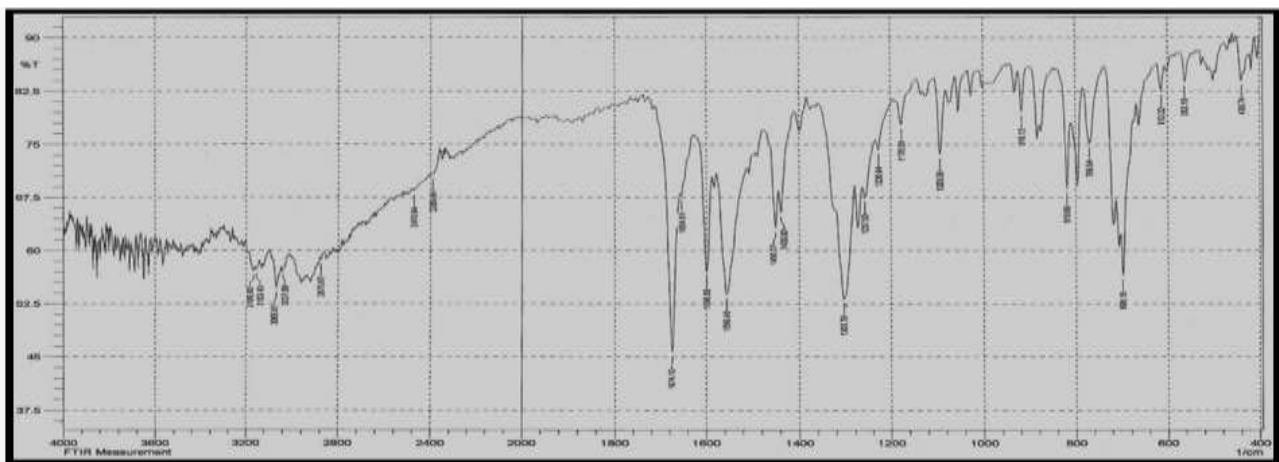


Figure 5: FTIR of poly decanoic acid with maleic anhydride with paracetamol

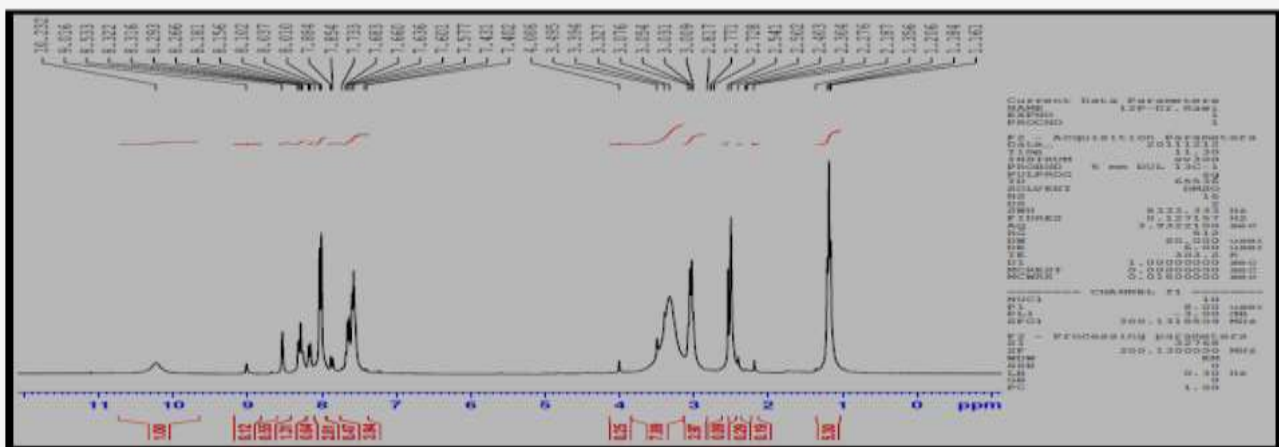


Figure 6: NMR of decanoic acid with thiourea

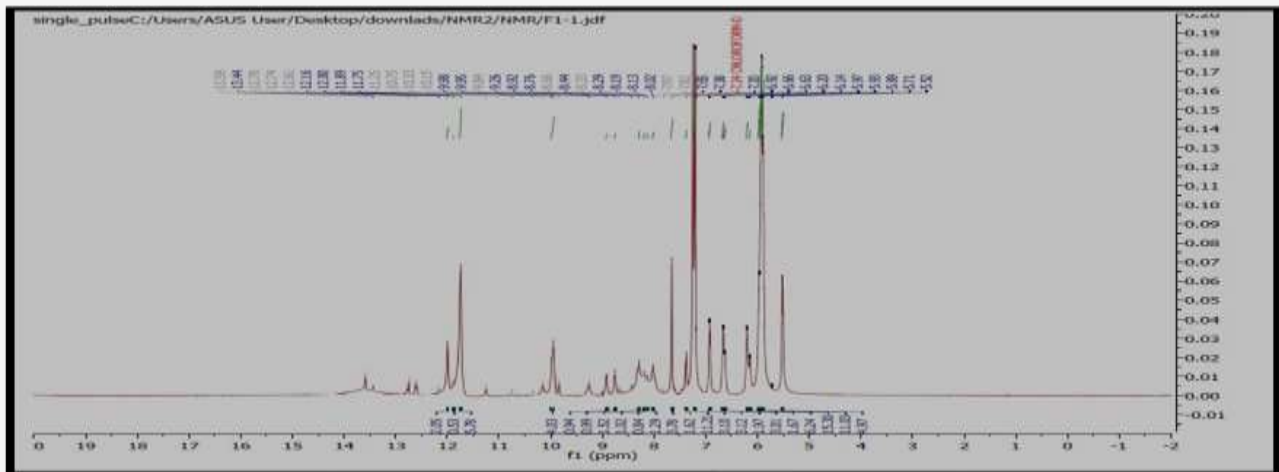


Figure 7: NMR of decanoic acid with thiourea with paracetamol

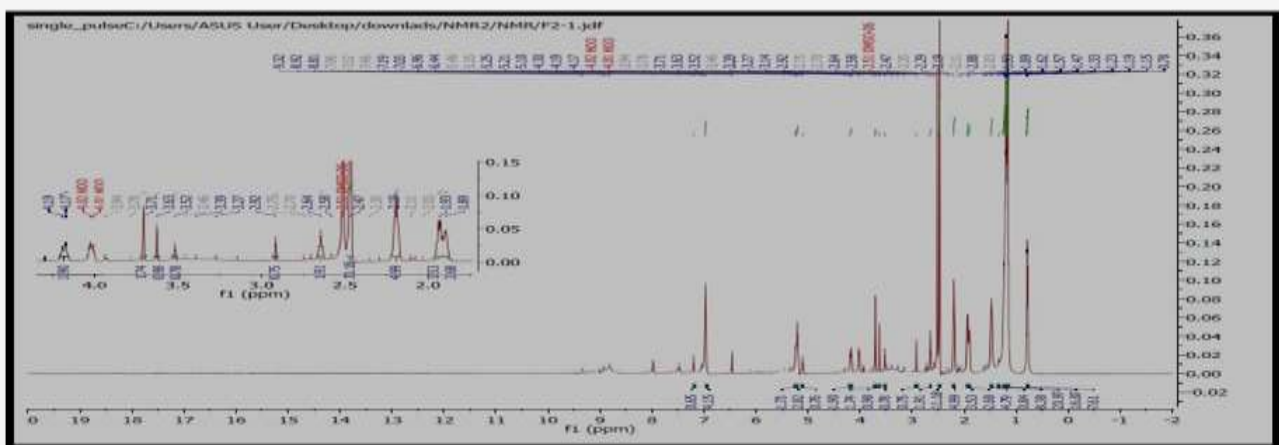


Figure 8: NMR of poly decanoic acid with maleic anhydride with paracetamol

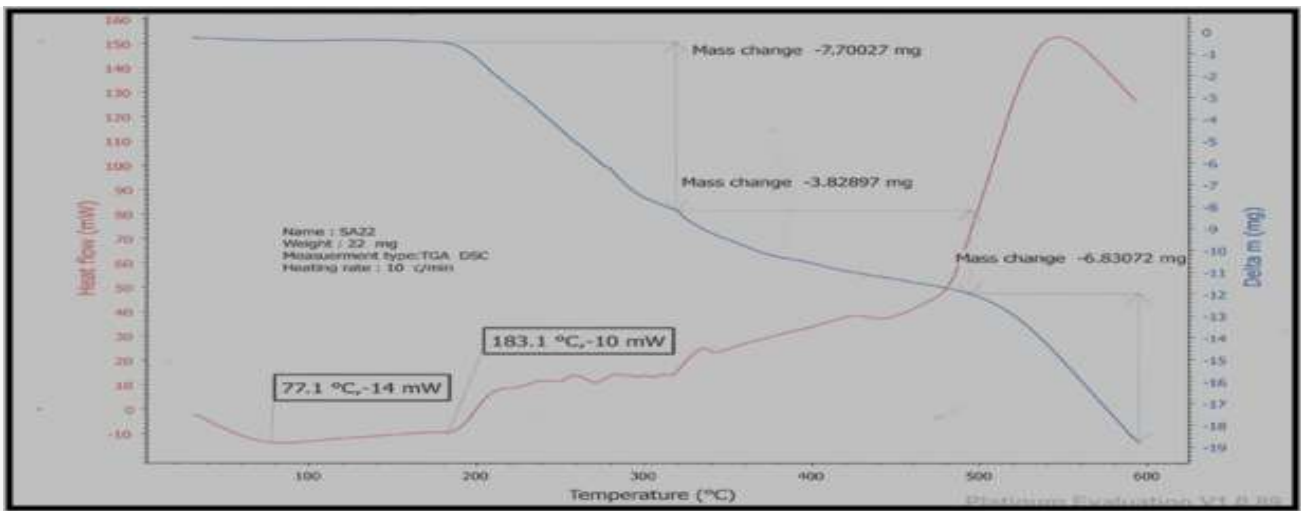


Figure 9: TGA of poly decanoic acid with maleic anhydride

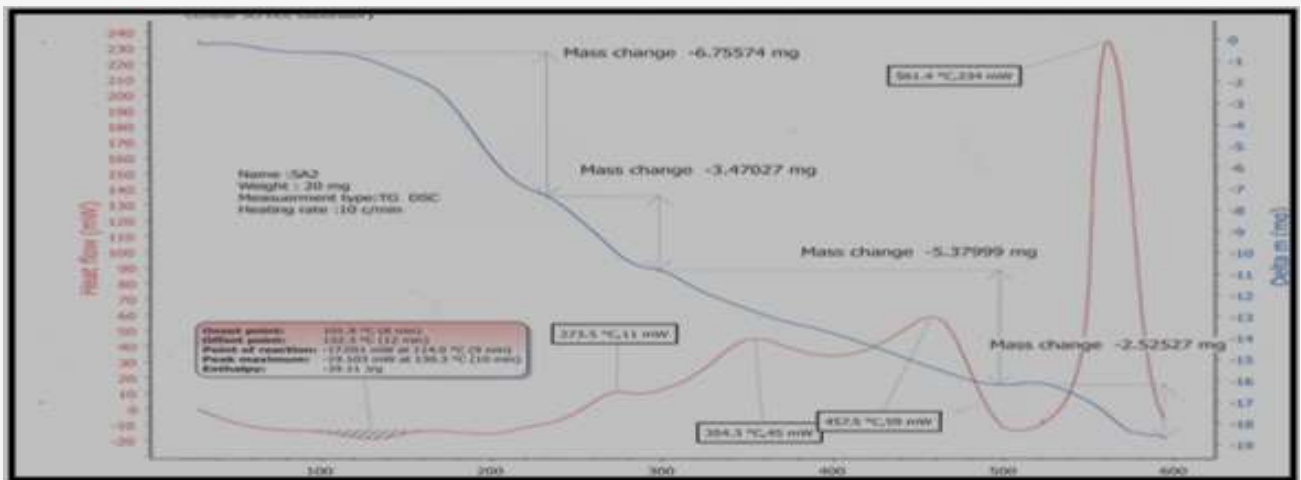


Figure 10: TGA of poly decanoic acid with maleic anhydride with paracetamol

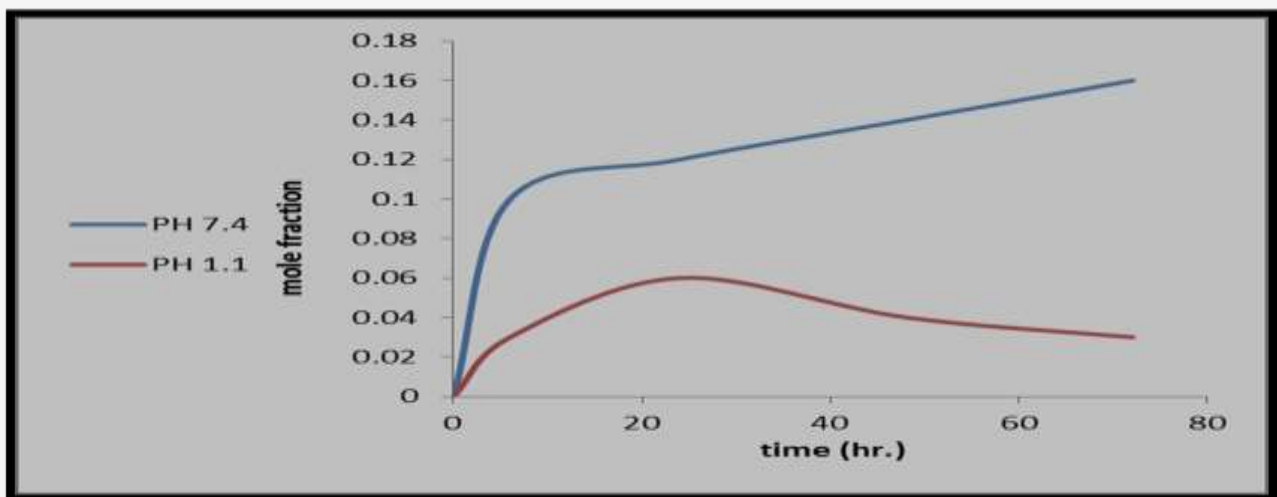


Figure 11: Drug release of poly decanoic acid with maleic anhydride with paracetamol in pH 1.1 and 7.4 at 37°C

Conclusions

This research complete synthesis and identification (IR and ¹H-NMR) of novel polymers by highly efficient methodology. This method depends on decanoic acid was extracted from coriander seed reacted with one groups of amin of thiourea; the 2nd group amin in thiourea was reacted with maleic anhydride a different drugs was used to

obtain novel derivatives. In addition to simplicity, this method is very fast and low cost for synthesis of these polymers.

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References

- Lalitha V, Kiran B, Raveesha KA (2011) Antifungal and antibacterial potentiality of six essential oils extracted from plant source, *Int. J. Eng. Sci. Technol.*, 3(4): 3029-3038.
- Potter TL, Fagerson IS (1990) Composition of coriander leaf volatiles, *J. Agric. Food Chem.*, 38: 2054-2056.
- Momin AH, Acharya SS, Gajjar AV (2012) *Coriandrum sativum*- review of advances in phytopharmacology. *IJPSR*, 3(5): 1233-1239.
- Iqbal-Bhanger M, Iqbal S, Anwar F, Imran M, Akhtar M, Zia-ul- Haq M (2008) Antioxidant potential of rice bran extracts and its effects on stabilization of cookies under ambient storage. *Int. J. Food Sci. Technol.*, 43(5): 779-786.
- Zoubiri S, Baaliouamer A (2010) Essential oil composition of *Coriandrum sativum* seed cultivated in Algeria as food grains protectant. *Food Chem.*, 122(4): 1226-1228.
- Eddouks M, Maghrani A, Lemhadri ML, Ouahidi L, Jouad H (2002) Ethnopharmacological survey of medicinal plants used for the treatment of diabetes mellitus, hypertension and cardiac diseases in the south-east region of Morocco (Tafilalet). *J. Ethnopharmacol.*, 82(2-3):97-103.
- Benyoussef EH, Saibi S (2013) Influence of essential oil composition on water distillation kinetics. *Flavour Fragrance J.*, 28(5):300-308.
- Msaada K, Hosni K, Ben Taarit M, Hammami M, Marzouk B (2009) Effects of growing region and maturity stages on oil yield and fatty acid composition of coriander (*Coriandrum sativum* L.) fruit. *Sci. Hortic.*, 120(4):525-531.
- Yap P, Yiap B, Ping H, Lim S (2014) Essential oils, a new horizon in combating bacterial antibiotic resistance. *Open Microbial. J.*, 8: 6-14.
- Egualé T, Tilahun G, Debella A, Feleke A, Makonnen E (2007) In vitro and in vivo anthelmintic activity of crude extracts of *Coriandrum sativum* against *Haemonchus contortus*, *J. Ethnopharmacol.*, 110(3):428-33.
- Saleh MA, Clark S, Woodard B, Deolu-Sobogun SA (2010) Antioxidant and free radical scavenging activities of essential oils. *Ethn. Dis.*, 20:S1-78-S182.
- Rajeshwari U, Shobha I, Andallu B(2011) Comparison of aniseeds and coriander seeds for antidiabetic, hypolipidemic and antioxidant activities. *Spatula DD*, 1(1):9-16.
- V Kolot, S Grinberg (2003) *J. Appl. Polym. Sci.* 91: 3838-3843.
- U Biermann, W Friedt, S Lang, W Lühs, G Machmüller, JO Metzger, MRG Klaas, HJ Schäfer, MP Schneider (2000) *Angew. Chem. Int. Ed.*, 39: 2206-2224.
- U Biermann, U Bornscheuer, MA Meier, JO Metzger, HJ Schafer (2011) *Angew. Chem. Int. Ed.* 50: 3854-3871.
- J Lu, S Khot, RP Wool (2005) *Polymer* 46: 71-80.
- H Bakhshi, H Yeganeh, S Mehdipour-Ataei, A Solouk, S Irani (2013) *Macromolecules* 46: 7777-7788.
- R Chen, JS Chen, C Zhang, MR Kessler (2015) *RSC Adv.*, 5: 1557-1563.
- G Guo, J Sun, C Zhao, Y Liu, C-M Liu (2016) *Green Chem.* 18: 1278-1286.
- Z Chen, BJ Chisholm, R Patani, JF Wu, S Fernando, K Jogodzinski, CD Webster (2010) *J. Coat. Technol. Res.*, 7: 603-613.
- SW Choi, D Wan Seo, Y Don Lim, Y Gi Jeong, MS Islam Mollah, H Park, T Whan Hong, W Gi Kim (2011) *J. Appl. Polym. Sci.* 121: 764-769.
- SN Khot, JJ Lascala, E Can, SS Morye, GI Williams, GR Palmese, SH Kusefoglou, RP Wool (2001) *J. Appl. Polym. Sci.*, 82: 703-723.
- A Guo, Y Cho, ZS Petrovic (2000) *J. Polym. Sci, Part A: Polym. Chem.* 38: 3900-3910.
- H Bakhshi, H Yeganeh, S Mehdipour-Ataei (2013) *J. Biomed. Mater. Res.*, A 101: 1599-1611.
- M Jalilian, H Yeganeh, MN Haghghi (2008) *Polym. Int.*, 57: 1385-1394.
- MSF Lie Ken Jie, MSK Syed-Rahmatullah (1992) *J. Am. Oil Chem. Soc.* 69: 359-362.
- M.S.F. Lie Ken Jie, Y.F. Zheng. *Chem. Phys. Lipids*, 49 (1988) 167-178.
- G Liu, X Kong, H Wan, S Narine (2008) *Biomacromolecules*, 9: 949-953.

28. R Slivniak, AJ Domb (2005) *Biomacromolecules*, 6: 1679-1688.
29. S Abraham, SS Narine (2009) *J. Polym. Sci, Part A: Polym. Chem.* 47: 6373-6387.
30. C Bartolini, L Mespouille, I Verbruggen, R Willem, P Dubois (2011) *Soft Matter.*, 7: 96-28.
31. S Tempelaar, L Mespouille, O Coulembier, P Dubois, AP Dove (2013) *Chem. Soc. Rev* 42: 1312-1336.
32. R Slivniak, A Ezra, AJ Domb (2006) *Pharm. Res.*, 23: 1306-1312.
33. F Suriano, O Coulembier, JL Hedrick, P Dubois (2011) *Polym. Chem.*, 2: 528-533.
34. MA Bertucci, SJ Lee, MR Gagne (2013) *Chem. Commun.*, 49: 2055-2057.
35. KS Bisht, YY Svirkin, LA Henderson, RA Gross (1977) *Macromolecules*, 30: 7735-7742.
36. JW Peeters, OV Leeuwen, ARA Palmans, EW Meijer (2005) *Macromolecules*, 38: 5587-5592.