

Synthesis of Few New Carrier Polymers Derived from 2-hydrazinylbenzo[d]thiazole

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ABSTRACT

2-hydrazinylbenzo[d]thiazole compound [1] is produced from reaction of 2-mercapto-benzothiazole with hydrazine hydride in ethanol. Compound [1] reacted with maleic anhydride in DMF to produce (Z)-4-(2-(benzo[d]thiazol-2-yl)hydrazinyl)-4-oxobut-2-enoic acid [compound (2)]. While the treatment of compound [2] with the ammonium persulfate ($(\text{NH}_4)_2\text{S}_2\text{O}_8$) (as the initiator) in order to produce compound [3], then compound [3] reacted with thionyl chloride in benzene to produce compound [4], finally compound [4] reaction with various drugs: cephalexin, amoxicillin, sulfamethizole, elecoxib obtained polymers [5–8]. The structure of synthesized compounds identified by spectral data: fourier transform infrared (FTIR) and proton nuclear magnetic resonance ($^1\text{H-NMR}$) spectroscopy. The polymers [5–8] have been screened for their antibacterial activities against *Staphylococcus aureus* (G+), *Escherichia coli* (G-) and compared with the drug (amoxicillin). The anticancer activity (Hep G2 (human liver cancer cell line) of some prepared polymers were also studied.

Keywords: 2-Hydrazinylbenzo[d]thiazole, Ammonium persulfate, Amoxicillin, Cephalexin, Hep G2.

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INTRODUCTION

Nitrogen- and sulfur-containing heterocycles can be found in a wide range of the natural products, commercially available drugs, agrochemicals, and compounds with the potential of becoming active pharmaceutical materials.¹⁻¹⁴ Which is why, there have been continued interests in the development of new approaches for synthesizing the biologically active fused heterocycles, which incorporate the benzo-thiazole fragment.¹⁵

2-mercapto-benzothiazole presents great interests in pharmacology and chemistry due to its wide range of biological activities.^{16,17} Benzothiazole must get the attention of the medicinal chemists due to their wide variety of biological activities, including anti-inflammatory, vasodilators, anti-tumor,¹⁸⁻²⁰ anti-tubercular, antifungal,²¹ antimicrobial,²² anticancer,²³ antidiabetic,²⁴ anti-convulsant,²⁵ antibacterial and antiviral activities.²⁶

Maleimide is a chemical compound with the formula $\text{H}_2\text{C}_2(\text{CO})_2\text{NH}$, maleimides can be defined as an important kind of heterocyclic compounds that exist in natural products²⁷ and find applications in organic as well as medicinal chemistry.²⁸ Maleimides have considerable significance in pharmaceutical products with biological particular, e.g., antibacterial, anticancer, anti-tumor, tuberculostatic activity, antimicrobial,

anti-viral and anti-genic activities.^{29,30} The purpose of this work was to synthesize new series schiff base with maleic anhydride by using a simple method of binding it with drug compounds and preparing new polymers using ammonium persulfate as a starting polymerizer.

EXPERIMENTAL

Materials

All of the chemicals have been supplied by Merck and Aldrich.

Methods

The fourier transform infrared (FTIR) spectra were recorded with the use of the KBr, discs on 8400 s Shimadzu spectrophotometer and FTIR spectrophotometer, Shimadzo (Ir prestige 21).

Proton nuclear magnetic resonance ($^1\text{H-NMR}$) spectra has been performed by: Bruker, ultra shield 400 MHz, University of Basra, Iraq, and have been reported in the ppm.

SYNTHETIC PROCEDURES (SCHEME 1)

1- Synthesis of 2-hydrazinylbenzo[d]thiazole [1]³¹

Solution of 2-hydrazinylbenzo[d]thiazole (1.67 g, 0.01 mol) in 5 mL ethanol, hydrazine hydrate (0.01 mol) was added

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Table 1: FT-IR of polymers [5-8]

Compound no.	ν (N-H)	ν (C-H)aroma.	ν (C-H) aliph.	ν (C=O) amide	ν (C=O)COOH	ν (C=N)
[5]	3213	3070	2933–2872	1654	1720	1620
[6]	3209	3055	2924–2816	1654	1716	1637
[7]	3214	3086	2912–2800	1674	-	1639
[8]	3190	3070	2947–2877	1687	-	1620

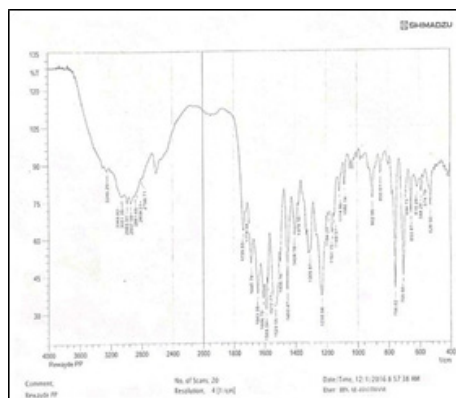
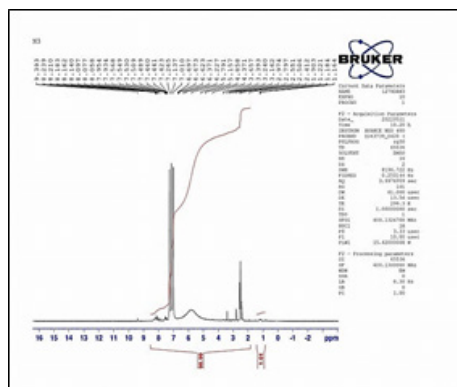
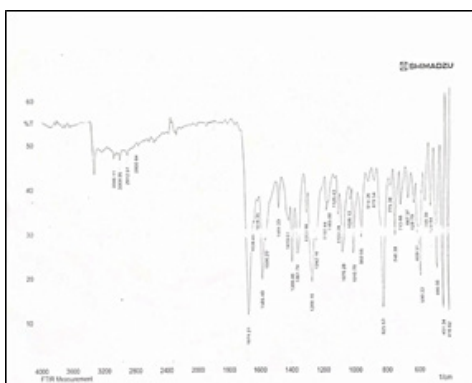
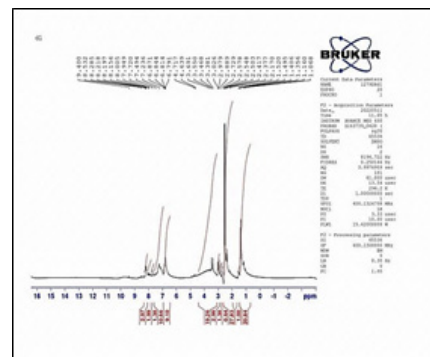
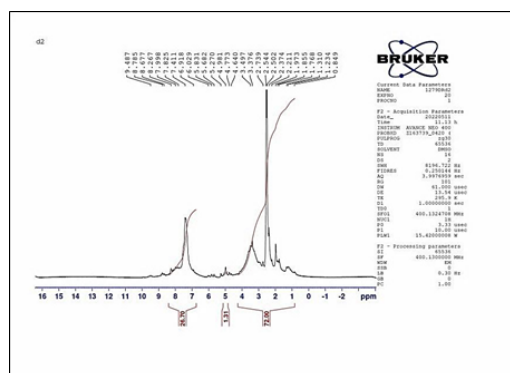

Figure 1: FT-IR of polymers [3]

Figure 2: ¹H-NMR of polymer [3]

Figure 3: FT-IR of polymer [7]

Figure 4: ¹H-NMR of polymer [7]

Table 2: The inhibition zone of some synthesized polymers.

Compound no.	<i>Escherichia coli</i>	<i>Staphylococcus aureus</i>
Amoxicillin	17	23
[3]	16	16
[6]	19	21
[7]	20	22
[8]	24	25

Table 3: The inhibition of cells growth of some synthesized polymers μ L/well

Compound no.	Inhibition of cells growth for Hep G2
[6]	44.7%
[7]	50.2%
[8]	59.6%


Figure 5: ¹H-NMR of polymer [8]

drop-wise with stirring and the mixt has been then refluxed for a duration of 24 hours, after that, the excess of solvent has been evaporated and the solid has been re-crystallized from the chloroform for the purpose of giving the desired off white product. yield 95%, mp, 93–95°C.

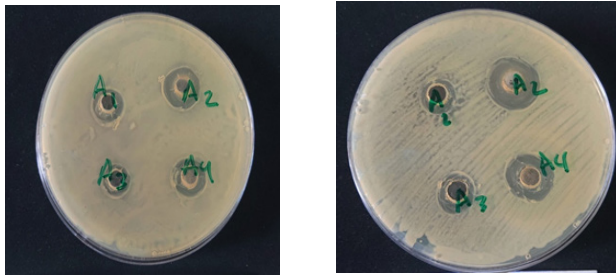


Figure 6: Antibacterial Activity of some polymers

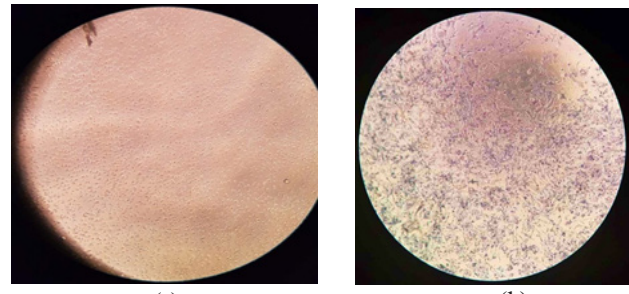
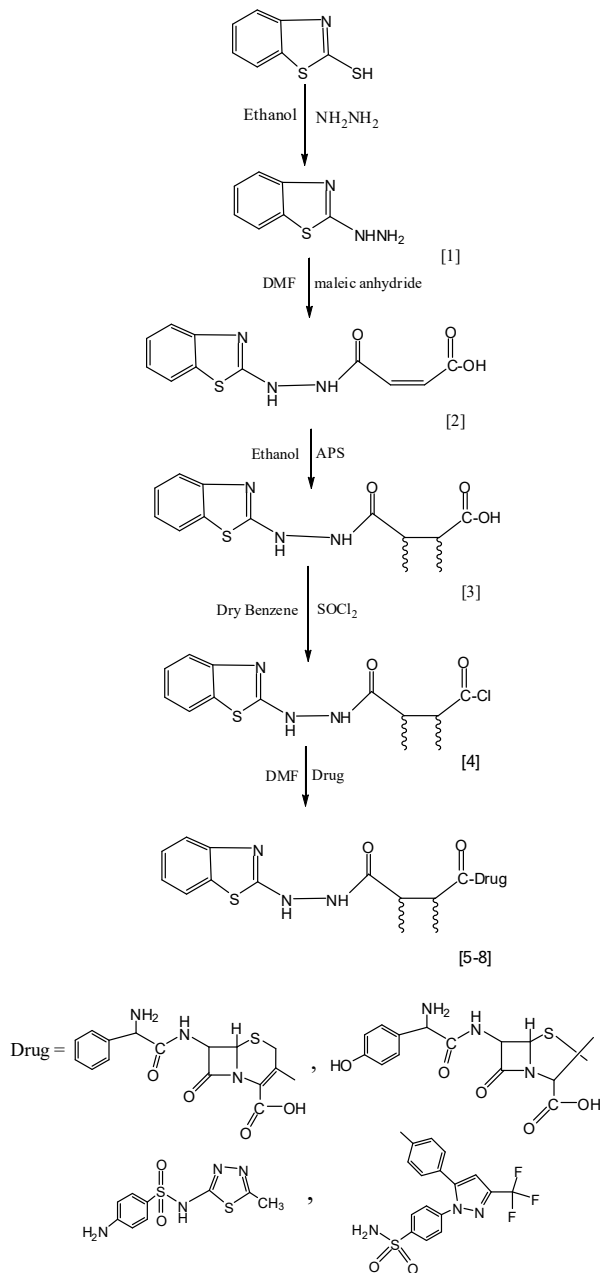


Figure 7: (a) Image of well Hep G2 before Staining (b) Image of well Hep G2 after Staining



Scheme 1: Synthetic procedures of polymers [1–8]

2-Synthesis of (Z)-4-(2-(benzo[d] thiazol-2-yl)hydrazinyl)-4-oxobut-2-enoicacid[2]³²

Mixing maleic anhydride (0.98 g., 0.01 mol) and (1.65 g., 0.01 mol) 2-hydrazinylbenzo [d]thiazole [1] in 25 mL of the DMF and after that, refluxed for approximately 4 hours, result has been washed by the diethyl ether and dried after that, at the temperature of the room (yield 75%).

3- Synthesis of Polymer [3]³³

(0.26 g, 0.001 mol) of the complex [2] has been mixed with (0.22 g) of the APS (*i.e.*, ammonium per sulfate) as a polymerization initiator in (15 mL) of the ethanol. This mix has been stirred for 2–3 hours at the temperature of the room, and after that, it was refluxed for approximately 12 hours after the filtering, it was washed by cold Et-OH absolute, dried and re-crystallized with the ethanol in order to give the required product [3].

3- Synthesis of Polymer [4]³⁴

0.01 mol of the compound [3] has been mixed with 0.010 mol of the thionyl chloride in 25 mL of the dry benzene then refluxed for (6 hours), SOCl_2 and benzene amount were separated after cooling under the vacuum.

4- Synthesis of Polymers [5-8]³⁵

A mix (0.010 mol) of compound [4] and (0.01 mol) of the drug (cephalexin or amoxicillin or sulfamethizole or celecoxib) in 20 mL. of the DMF. This mix has been refluxed for approximately 3 hours after that filtering, it has been washed by absolute ethanol, dried then re-crystallized from the Et-OH for producing the needed product [5-8].

RESULT AND DISCUSSION

Compound [1] has been synthesized *via* reacting 2-mercapto-benzothiazol with hydrazine hydride in ethanol. FTIR of compound [1] display the appearance bands at (3396 cm^{-1} and 3278 cm^{-1}) have been respectively a result of the asymmetrical and symmetrical stretching vibrations of the ($-\text{NH}_2$) group and disappearance (SH) group at 2357 cm^{-1} .

Compound [2] produce from reaction compound [1] with maleic anhydrides in DMF. Compound [2] was identified by FTIR spectroscopy. The FTIR absorption (ν, cm^{-1}): 3180 (NH) group, 3039 (C-H arom.), 3400-2400 (OH), 1693 (C=O) of the carboxylic acid, 1630 (C=N), 1660 (C=O-NH), 1195 (C=S) and 1591(C=C) Aromatic.³⁶ Polymer [3] was synthesized by the

reaction compound [2] with ethanol by the use of the APS as the initiator. FTIR spectrum (ν , cm^{-1}) of polymer [3], Figure 1: show stretching band that refers to O-H of COOH moiety in a region (3400–2400) cm^{-1} , a stretching band of N-H group had appeared at 3246 cm^{-1} , 3051 (C-H aromatic), 2937,2881 (C-H aliph.) and a stretching band to C=O for the COOH had appeared at 1685 cm^{-1} . ^1H NMR (δ ppm) of polymer [3], Figure 2 show: signals at 1.22 ppm due to six protons for two CH_3 groups, -CH=CH chemical shifting disappears and show chemical shifting at δ (2.79-3.39) because of -[CH-CH] $_n$ -, (7.26-7.56) ppm that attributed to the four aromatic protons and 5.79 ppm, 8.23 ppm and 9.39 ppm for a sharp signals for a single proton could be a result of to CH=N, NH and one proton of (OH) carboxylic group. Polymer [4] through reaction polymer [3] with thionyl chloride in the dry benzene, FTIR for polymer [4], had shown band vanishing at (3400–2400) cm^{-1} due to (OH) group of the carboxylic acid in addition to band appearance at (1770) cm^{-1} that had been connected to acyl chloride. Polymers [5-8] were synthesized from reacted polymer [4] with drugs (cephalexin, amoxicillin, sulfamethizole, celecoxib). FTIR spectrum for polymer [7], Figure 3, the disappearance of absorption band to acyl chloride and absorption band appearance at (1674), (1639), (1076) cm^{-1} due to C=O-NH, C=N and S=O, respectively.³⁷

^1H NMR (δ ppm) of the polymers [7], Figure 4: singlet signal at δ 1.06 refer to (CH_3) group that is related to the ring of 1,3,4-thiadiazole, a sharp signals at 1.16 ppm due to six protons for two CH_3 groups, -CH=CH chemical shifting has disappeared and shown chemical shifting at δ (2.88–2.97) because of -[CH-CH] $_n$ -, multiple signals at δ (6.79–8.21) that refer to aromatic protons, singlet signal at δ 4.71 as a result proton of NHNH-C=O , signal at δ (8.53) attributed to the one proton of NHNH-C=O and signal at δ (9.40) attributed to NH-S=O

^1H NMR (δ ppm) for polymer [8], Figure 5: signals at δ (0.84) and (1.13) ppm due to proton of (CH_3)-ph and $[(\text{CH}_3)]_2$ group respectively, -CH=CH chemical shifting is disappear and show chemical shifting at δ 3.37 ppm because of -[CH-CH] $_n$ -, signal at δ 4.77 indicate proton of NH-NH C=O , signal at δ 5.83 refer to the proton of a ring, multiple signals at (δ 6.91- δ 8.26) that attributed to aromatic protons, signals at δ (8.78) and (9.48) ppm attributed to the NH-C=O and NH-S=O , respectively. FTIR of all polymers is listed in Table 1.

Biological Activity³⁸

The polymers [3,6,7,8] have been tested for antibacterial activities towards *S. aureus* (G+) and *E. coli* (G-) *in-vitro* through agar well diffusion approach (Barry, 1977), a standard medication that has been utilized for the purpose of comparing with synthesized polymers is amoxicillin (50 $\mu\text{g}/\text{mL}$). Results have shown that all of the polymers had a higher diameter of the zone of growth inhibition, which has been given in Table 3. Compound [8] had shown good inhibition towards *E. coli*; this could be related to the presence of the drug because this drug is antibacterial that is obtained from a sulfonamide. It is effective as an antibiotic against broad range of the G- and G+ bacteria.

Also, polymer [8] contains maleimide group and schiff base materials which have good activities against bacteria. Some polymers and their antibacterial activities listed in the Figure 6.

Anticancer Activity

The polymers [6-8] have been screened for anticancer activities, utilizing one cancer cell line type: Hep G2 (human liver cancer cell line). Freshney's protocol for the cell culture media, solutions and reagents have been followed.³⁹ HepG-2, viability after the addition of various concentrations of compounds [6-8] has been determined through the use of an enzyme-linked immunosorbent assay (ELISA) reader at 575 nm wavelength (Figure 6).

Cell growth-inhibiting rate has been calculated as follows⁴⁰:

$$\text{Rate of Inhibition} = \frac{\text{mean of control} - \text{mean of treatment}}{\text{mean of control}} \times 100$$

Polymer [8] showed more than 50% inhibition for Hep G2 (human liver cancer cell line).

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